

Periodic Table of the Elements

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1 IA		2 IIA																2 He
1 H	2 Be																	1 1
3 Li	4 Be																	2 2
11 Na	12 Mg	3 IIIIV	4 IVB	5 VB	6 VIB	7 VIIIB	8 VII	9 IB	10 IIB	11 IB	12 IIB	13 Al	14 Si	15 P	16 S	17 Cl	18 Ar	3 3
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr	4 4
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe	5 5
55 Cs	56 Ba	57-71 Hf	72 Ta	73 W	74 Re	75 Os	76 Ir	77 Pt	78 Au	79 Hg	80 Tl	81 Pb	82 Bi	83 Po	84 At	85 Rn	6 6	
87 Fr	88 Ra	89-103 Rf	104 Db	105 Sg	106 Bh	107 Hs	108 Mt	109 Ds	110 Rg	111 Cn	112 Uut	113 Fl	114 Uup	115 Lv	116 Uus	117 Uuo	7 7	
■ Alkali Metals ■ Alkali Earth Metals ■ Transition Metals ■ Other Metals ■ Metalloids ■ Other Non Metals ■ Halogens ■ Noble Gases ■ Lanthanides & Actinides																		

This is one of many variations by different authors, depending on their personal preferences. Despite such variations, the one above is a good representative of the generally accepted Periodic Table. However, in this model, zinc is located in the yellow s block of transition metals, rather than the orange d block of post-transition metals. This is still a area of debate among chemists.

